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Acids And Bases

Chapter 19

An acid is a substance

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that contains hydrogen
and ionizes to produce
hydrogen ions in

aqueous solution; A
base is a substance that
contains a hydroxide
group and dissociates to
produce a hydroxide ion
in aqueous solution.

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Section 19.3 □ Strengths

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Acids And

of Acids and Bases In

19 Answers
In general, strong acids completely dissociate in aqueous solution. Weak acids only slightly ionize in aqueous solution. Strong acids include HCl, HNO₃, and H₂SO₄.

Chapter 19 □ Acids,
Bases, and Salts

Acids are hydrogen
containing compounds,

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Acids And

ionize to yield hydrogen

ions in aqueous

solutions. Bases are

compounds that ionize

to yield hydroxide ions

in aqueous solutions.

Monoprotic acids. acids

that contain 1 ionizable

hydrogen such as nitric

acid (HNO_3) diprotic

acids.

Chemistry Chapter 19:

Bases and Acids

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accept a positive hydrogen ion. Thus, an acid-base reaction in the reverse direction can occur. The acid H_3O^+ can react with the base X^- to form water and HX and the following equilibrium is established.

598 Chapter 19 Acids and Bases

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Acids And

HX(aq) + H₂O(l) ⇌ H₃O⁺(aq) + X⁻(aq)

Acid Base

Conjugate acid

Conjugate base

Chapter 19: Acids and
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and salts. STUDY.

PLAY. how did

Arrhenius define an acid
and a base? according to

Arrhenius, acids are

hydrogen-containing

compounds that ionise

to yield hydrogen ions

(H^+) in aqueous

solution bases are

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Acids And

compounds that ionise
to yield hydroxide ions
(OH⁻) in aqueous
solution.

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Chemistry Page

Brønsted-Lowry acids
and bases come in pairs.

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Acids And

A conjugate base is the remainder of the B-L acid, after it donates its H^+ . A conjugate acid is the substance formed when the B-L base gains a H^+ . Thus, a conjugate acid-base pair is related by the . loss or gain. of a single hydrogen ion.

Chapter 19 Acids,
Bases, and Salts

Chapter 19 Acids,
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Acids And

Bases, and Salts

What are the properties of acids and bases? acids taste sour, will change the color of an acid-base indicator, and can be strong or weak electrolytes in Samples

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Section 19.1 Acid-Base

Theories

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Acids And

OBJECTIVES:

1. Compare and contrast

acids and bases as

defined by the theories

of: a) Arrhenius, b)

Brønsted-Lowry, and c)

Lewis.

4.

Chemistry - Chp 19 -

Acids, Bases, and Salt -

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Chapter 19: Acids and

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Acids And

Bases You're probably already aware from what you've learned in other classes that acids and bases are all over the place. From the lemon you put in iced tea to the Drano put down the drain to keep your hair from clogging the drain, acids and bases are an important part of our lives.

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