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Answer the following questions, circle the best answer. 1) Which of the following are isotopes of

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Answers Chapter 5
each other? a) ^{14}C and ^{14}N
b) ^3H and ^4He c) ^2H and ^1H
d) none of these . 2) The net charge on an atom that has 13 protons, 12 neutrons, and 10 electrons is . a)

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Atomic Structure Answers 1.

In the nuclear model of the atom, protons (and neutrons) are housed in a small, dense nucleus. Electrons surround

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the nucleus in an area of mostly empty space. 2. If electrons are electrically attracted to nucleus and would, therefore, have centripetal acceleration in order to orbit the nucleus.

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ATOMIC STRUCTURE AND THE PERIODIC TABLE CHAPTER 4 WORKSHEET PART A Given the following isotopes, determine the atomic number, the mass number, the number

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Answers Chapter 5
of protons, electrons and
neutrons. Isotope Symbol
131i S31 Us |He SAI 81W 36Kr
S1DU 37KD Atomic Number Mass
Number Protons Electrons
Neutrons Isotope Name PARTB

~~|He SAI~~

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Structure Worksheet 5 Answer

Key Measuring Atomic Mass

Instead of grams, the unit

we use is the Atomic Mass

Unit (amu) It is defined as

one-twelfth the mass of a

carbon-12 atom. Carbon-12

chosen because of its

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isotope purity. Each isotope has its own atomic mass, thus we determine the average from percent abundance. Page 2/11

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