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your answers). (a)

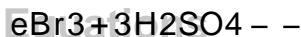
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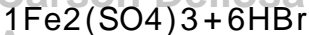
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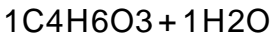
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Tip # 1: When you are trying to balance the chemical equations, you should remember that you

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can only change the value of coefficient in front of the element or compound, and not the subscript.

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Answer Key

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Operations with

Real Numbers

Adding Real

Numbers $-7 + 6 =$

-1 Add. 1. $2.7 +$

$(-4.8) = 2. 1.45 +$

$2.65 + (-9.43) = 3.$

$-55 + (-8) + (-4)$

$+ 54 = 4. 3.54 +$

$4.27 + 7.43 = 5. 10$

$+ 7 + (-7) + (-10)$

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$$\begin{aligned} &= 6 \cdot 16 + 21 + \\ &(-3) + 7 = 7 \cdot 10 + \\ &7 + (-16) + 9 + \\ &(-30) = 8. \end{aligned}$$

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The following
atoms all undergo
alpha particle
emission. Write the
complete nuclear
equation. alpha
particle + Pb-206

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alpha particle +

Th-234 → alpha

particle + Ra-234

alpha particle +

Po-218. The

following atoms all
undergo beta decay.

Write the complete
nuclear equation.

beta particle (e⁻) +

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Equations: Enter an equation of a chemical reaction and click 'Balance'.

The answer will appear below;

Always use the upper case for the first character in the element name and the lower case for the second character.

Examples: Fe, Au,

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Co, Br, C, O, N, F.

Compare: Co -
cobalt and CO -
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Always solve equations in this

order: 1. Perform operations inside parentheses. 2.

Simplify exponents.

3. Multiply and divide from left to right. 4. Add and

subtract from left to right. Property of Equality If you add,

subtract, multiply,

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or divide both sides of a true equation by the same amount, the equation remains true. Commutative Property of Addition

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nitrate react to
form Zinc Nitrate
and Lead. $Zn +$

$Pb(NO_3)_2 \rightarrow Zn(NO_3)_2 + Pb$ Single
Replacement 2.

Aluminum Bromide
and Chlorine gas
react to form

Aluminum Chloride
and Bromine gas.

$2AlBr_3 + 3Cl_2$

$2AlCl_3 + 3Br_2$

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nitrate react ...

Chemical reactions

have the same

number of atoms

before the reaction

as after the

reaction. Balancing

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is a basic skill in chemistry and testing yourself helps retain important information. This collection of ten chemistry test questions will give you practice in how to balance chemical reactions.

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numbers that you

have entered into

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