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chemists use balanced chemical equations? what does a balanced
chemical equation? what is stoichiometry. in terms of what
quantities can you interpret as a basis to calculate how much
reactant is needed or how much? what amounts of reactants to
mix and what amount of product to? the calculation of
quantities in chemical reactions. numbers of atoms ...

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Chapter 12 Stoichiometry 127 SECTION 12.1 THE
ARITHMETIC OF EQUATIONS (pages 353-358) This section
explains how to calculate the amount of reactants required or
product formed in a nonchemical process. It teaches you how
to interpret chemical equations in terms of interacting moles,
representative particles, masses, and gas volume at STP.

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