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~~CHAPTER 11: STOICHIOMETRY~~

Chapter 12 Stoichiometry Section 12.1 The Arithmetic of Equations Using Balanced Chemical Equations Chemists use balanced chemical equations as a basis to calculate how much reactant is needed or product is formed in a reaction. Stoichiometry = calculation of quantities in chemical reactions is a subject of chemistry.

~~Chemistry Chapter 12 Stoichiometry Section 12.1 The ...~~

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CHAPTER 9 REVIEW Stoichiometry SECTION 2 PROBLEMS Write the answer on the line to the left Stoichiometry worksheet answers chapter 9. Show all your work in the space provided. 1. 4 Stoichiometry worksheet answers chapter 9. 5 mol The following equation represents a laboratory preparation for oxygen gas: 2KClO 3(s) ? 2KCl(s) 3O 2(g) How many moles of O 2 form if 3. 0 mol of KClO 3 are totally ...

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