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### 22CHAPTER Study Guide Key

#### Concepts 22.1 Hydrocarbons •

Because carbon has four valence electrons, carbon atoms always form four covalent bonds. • The carbon atoms in an alkane can be arranged in a straight chain or in a chain that has branches. • Molecules of hydrocarbons, such as alkanes, are nonpolar molecules. 22.2 Unsaturated Hydrocarbons

### CHAPTER 22 Study Guide - Quia

A controlled process by which hydrocarbons are broken down or

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rearranged into smaller, more useful molecules natural gas This is typically made up of 80% methane, 10% ethane, 4% propane, and 2% butane

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718 Chapter 22 22CHAPTER Study Guide Key Concepts 22.1

Hydrocarbons • Because carbon has four valence electrons, carbon atoms always form four covalent bonds. •

The carbon atoms in an alkane can be arranged in a straight chain or in a chain that has branches.

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Chapter 22: Substituted Hydrocarbons and Their Reactions. 784 Formic acid. BIGIdeaThe substitution of different functional groups for hydrogen atoms

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in hydrocarbons results in a diverse group of organic compounds. 22.1

Alkyl Halides and Aryl Halides.

MAIN Idea A halogen atom can replace a hydrogen atom in some hydrocarbons.

## Chapter 22: Substituted Hydrocarbons and Their Reactions

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ISBN-13: 978-0-13252-576-3,  
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## Chapter 22 - Hydrocarbon Compounds - 22.1 Hydrocarbons ...

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Answers Hydrocarbons are organic  
molecules that only contain two  
elements: carbon and hydrogen.  
Common hydrocarbons include  
methane, propane, and butane.  
Hydrocarbons are the primary  
constituents of fossil...

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optical isomers. Complete the concept map by writing the term below that fits the description at the right of each box.

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