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a) Ethane in staggered conformation has C 3-rotational axis passing through the C-C bond.It has three C 2-rotational axes perpendicular to the principal C 3-rotational axis and through the mid-point of the C-C bond.It has not any h-plane.But it has 3 d-planes containing two carbons and anti two hydrogens.Hence the point group of ethane in the staggered conformation will be D 3d.

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chemistry syllabus. This chapter touches on several fundamental concepts in the field of chemistry (such as hybridization and the modern theories on chemical bonding). The concepts covered in this chapter are listed below. VSEPR theory

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