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MULTIPLE CHOICE

On the line at the
left of each

statement, write

the letter of the

choice that best

completes the

statement or

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Answers the
question.

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ANSWER Answer
the following
questions in the

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space provided. 1.

a Technological development of a chemical product

often (a) lags

behind basic

research on the

same substance.

(b) does not

involve chance

discoveries. (c) is

driven by curiosity.

(d) is done for the

sake of learning

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Gravity. Created
by. dolphinking.
study the thing.

Terms in this set
(25) Chemical
formula. indicates

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the relative
number of atoms of
each kind in a
chemical

compound. Reveals
the number of
atoms of each
element contained
in a single
molecule ...

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The Periodic Law

SECTION 1 SHORT

ANSWERS Answer

the following

questions in the

space provided. 1.

c In the modern

periodic table,

elements are

ordered (a)

according to

decreasing atomic

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mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

5 The Periodic Law

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Properties SECTION

1 SHORT ANSWER

Answer the

following questions

in the space

provided. 1. Use

the guidelines in

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Table 1 on page
437 of the text to
predict the
solubility of the
following

compounds in
water: soluble a.
magnesium nitrate
insoluble b. barium
sulfate insoluble c.
calcium ...

13 Ions in Aqueous
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Chemical Bonding

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ANSWER Answer

the following

questions in the

space provided. 1.

a A chemical bond

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between atoms
results from the
attraction between
the valence

electrons and of
different atoms. (a)
nuclei (c) isotopes
(b) inner electrons
(d) Lewis structures

2. b A covalent
bond consists of (a)
a shared electron.

6 Chemical

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CHAPTER 9 REVIEW

Stoichiometry

MIXED REVIEW

SHORT ANSWER

Answer the following questions in the space provided. 1. Given

the following equation: $C_3H_4(g)$

$+ xO_2(g) \rightarrow 3CO$

$2(g) + 2H_2O(g)$ 4

a. What is the

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value of the
coefficient x in this
equation? 40.07
g/mol b. What is
the molar mass of
 C_3H_4 ? 2 mol O
2:1 mol H 20 c.
What is the mole
ratio ...

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