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Advantage - equation: $M.A. =$
_____ The greater the M.A.
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to provide standard rules
for writing and ...

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...

Assuming the pressure stays constant, at what temperature will the volume of the gas be 604 cm³?

Given: $V_1 = 506 \text{ cm}^3$. $T_1 = 147 \text{ }^\circ\text{C}$. $V_2 = 604 \text{ cm}^3$. $T_2 = ?$

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$V_1/T_1 = V_2/T_2$. $506 \text{ cm}^3/420 \text{ K}$
 $= 604 \text{ cm}^3/T_2$. $T_2 = 501 \text{ K}$.
or. $501 - 273 = 228 \text{ }^\circ\text{C}$.

Remember to convert
temperature to Kelvin (420
K) Episode 902

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