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~~to Buffer Solutions~~

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~~Calculations,~~

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~~Acid-Base~~

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~~# 8 Acid Base~~

~~Neutralization~~

~~Reactions \u0026~~

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~~Problems, pH~~
~~Calculations Given~~
 ~~K_{a1} , K_{a2} \u0026amp;~~
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~~Base Titration~~

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~~Calculations, Weak~~

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~~Equivalence Point,~~

~~Chemistry~~

~~Problems~~ How to

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buffer solution ~~Acid~~

~~Base Titration~~

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Chemistry

Buffer solutions are

a mixture of

substances that

have a fairly

constant pH

regardless of

addition of acid or

base. They are

used in medicine,

industry and

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manufacturing to
keep the pH of a
solution stable
regard- less of
other reactions that
might be occurring.

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buffer system in the

bloodstream of

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Answer is the

carbonic
acid/bicarbonate

buffer. $\text{H}_2\text{CO}_3 +$
 $\text{H}_2\text{O} \rightleftharpoons \text{HCO}_3^- +$

H_3O^+ + The

carbonic acid in the
bloodstream is

formed by the
combination of

carbon ...

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system in the
bloodstream of
animals is ...

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Chemistry

Buffer solutions are a mixture of substances that have a fairly constant pH regardless of addition of acid or base. They are used in medicine, industry and manufacturing to keep the pH of a

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solution stable
regardless of other
reactions that might
be occurring.

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Chemistry-Modified

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one keeps adding

1.0 M HCl to this

buffer until a total

of 101 mL of 1.0M

HCl have been

added. At this point,

the pH of ...

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Answers Buffers

Answer keys to some hw problems can be downloaded below. Read chapter 18 notes on batteries, go through powerpoint and answer the following questions in the textbook pg 879 Q's 39-59 odd: 4/21 reviewed hw

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17 POGIL activity
on chapter 17 and
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sections 18.1 and
18.2. Look through
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