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object has because of its motion. kinetic
theory. All matter consists of tiny particles
that are in constant motion.
molecules/atom. the particles in a gas
usually consist of. insignificant. The

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particles in a gas are considered to be small, hard spheres with an _____ volume. empty space ...

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theory . All matter consists of tiny particles
that are in constant motion.
molecules/atom. the particles in a gas
usually consist of. empty space. what is

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between particles in a gas? rapid, constant, and random. the motion of the particles in a ...

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atmospheric pressure. Allotrope. Two or more ...

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kinetic energy. kinetic theory.

molecules/atom. insignificant. The energy
an object has because of its motion. All

matter consists of tiny particles that are in
constant mot... the particles in a gas

usually consist of. The particles in a gas

are considered to be small, hard sphere...

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STUDY. PLAY. Kinetic Energy. The energy an object has because of its motion. Kinetic Theory . The theory that states that all matter consists of tiny particles that are in constant motion. When applied to gases makes the following fundamental assumptions: The particles in a gas are small hard spheres with insignificant

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volume, The motion of gas particles is ...

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it's important for you to begin having that
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155. Section Review. Objectives. $\hat{\text{€}} \phi$

Determine the number of valence ... €^1

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acquire the stable electron structure of a

noble gas by losing electrons. ... Answer

the following in the ...

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