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Dissolving: Dissociation,

Dispersion or Ionization

Substances can dissolve

in water by three ways:

1. dissociation 2.

dispersion 3. ionization

Dissociation is the

process by which an

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Physical compound

separates into ions Ex:

NaCl Dispersion is the

process by which a

compound breaks into

small pieces that spread

throughout the water

Ex: sugar dissolves in

water Ionization is the

process by which neutral

molecules gain or lose

electrons (forming ions)

Ex: hydrogen chloride

gas & water react to ...

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Chapter 4 Atomic

Structure Atom V

Element ATOMIC

THEORY Democritus

(4th century B.C) said

the universe was made

of invisible units called

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atoms In 1808, an Englishman named John Dalton proposed a theory: 1. every element is made of tiny unique particles called atoms that cannot be subdivided 2. atoms of the same element have the same ...

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Solids have a definite

shape and volume

because particles in a

solid vibrate about a

fixed point

Adding/taking Energy

out of a system When

you add energy to a

system the particles

begin to move faster As

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they move faster they collide and create friction (Heat) As they continue we get a phase change Six Common Phase Changes phase change: is the reversible physical change that occurs when a substance changes from one state of matter to another Six common phase changes: melting, freezing ...

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into the world around
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of living things,
including plants,
animals, and mic.... the
study of the properties
of matter and how
matter changes.

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